

*Tetrahedron Letrers, Vol. 38. No. 23, pp. 4103-4106,* **1997 0 1997 Elsevier Science Ltd All tights reserved. Printed** in Great **Britain**  PII: SOO40-4039(97)00838- 1 *0040-4039/97 \$17.00 + 0.00* 

## **Highly Substituted Enyne Formation by Coupling Reaction of Alkenylzirconium Compounds with Alkynyl Halides**

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Abstract: Alkenylzirconium compounds, which were prepared by carbozirconation or hydrozirconation **of alkynes. reacted with alkynyl halides in the presence of CuCl to give highly substituted 1,3-enynes in good yields. This reaction was done conveniently in one-pot from alkynes. 0 1997** Elsevier Science Ltd.

Conjugated enynes are of great interest in organic synthesis. 1.2 Recent organometallic chemistry has demonstrated stereoselective formation of 1,3-enynes.3 Two major coupling patterns are shown in Scheme 1 which involves i) a coupling of an alkynyl metal **1** with an alkenyl halide 2 (Type *A)* and, conversely, ii) a coupling of an alkynyl halide 3 with an alkenyl metal 4 (Type B) to give 1,3-enynes 5. However, the crucial problem lies in the limitation of general stereoselective synthetic methods of highly substituted alkenyl compounds 2 or 4.



We have recently reported the formation of alkenylzirconium compounds (4 where M=Zr,  $R^3 \neq H$ ) by carbozirconation reactions,<sup>4,5</sup> and also the coupling reaction between the  $sp<sup>2</sup>$  carbon centers of 4 and aryl or alkenyl iodides mediated by CuCl with a catalytic amount of  $Pd(PPh<sub>3</sub>)<sub>4</sub>$ .6 We would like to report herein the coupling reaction of alkenylzirconium compounds with alkynyl halides which gives *highly* substituted enynes 5 according to the Type B pattern.

Phenylethynyl bromide reacted with alkenylzirconium compound 6, prepared by hydrozirconation of lhexyne, in the presence of CuCl to give disubstituted enyne 7 in 87% yield (eq. 1).



For preparation of *trisubstituted* enynes where  $R^1 = H$ , carbocupration<sup>7</sup> or carboalumination<sup>8</sup> of terminal alkynes gave alkenylmetal species 4, and the subsequent coupling reaction with alkynyl halides was successful. However, because of the limited examples of carbometallation of internal alkynes,<sup>1b</sup> the preparation of *tetrasubstituted enynes by coupling between sp*<sup>2</sup>- and sp-carbons has not been systematically studied, to the best of our knowledge.



Alkenylzirconium compounds 8a and 8b were prepared from alcoholysis of zirconacyclopentenes by t-BuOH," and **9a** and **9b** were prepared by the reaction of zirconacyclopentenes with diallyl ether.5a.b Dienylzirconium compound 10 was prepared from a zirconacyclopentadiene by the treatment of *t*-BuOH for 12 h at room temperature. We found that alkenylzirconium complexes of types 8,9 and 10 efficiently reacted with alkynyl halides in the presence of CuCl to give fully substituted 1,3-enynes  $5 (R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup> \neq H)$  in a wellstereodefined manner (eq. 2). The reaction proceeded in the absence of a palladium catalyst in contrast to the arylation reaction.<sup>6,7</sup> When alkynyl iodides were employed, the products were contaminated typically with  $10-$ 20% of alkenyl iodides **11,** which were not readily separable. On the other hand, the use of alkynyl bromide afforded enynes in high yields. Formation of alkenyl bromides was not observed.

$$
R^{1} \longrightarrow R^{2} \longrightarrow \text{CP2Z1} \longrightarrow \text{CP2Z1} \longrightarrow \text{CP2Z1} \longrightarrow \text{CP2Z1} \longrightarrow \text{CP3} \longrightarrow \text{CP4} \longrightarrow \text{CP4} \longrightarrow \text{CP5} \longrightarrow \text{CP6} \longrightarrow \text{
$$

As shown in Table 1, the yields were good to high. In all cases, alkenylalkoxyzirconium compounds were treated by an equimolar amount of CuCl and alkynyl bromide at room temperature for 3 h. The reactions were carried out under an inert gas atmosphere, and the reaction mixtures were analyzed by GC. All the products were purified by flash column chromatography on silica gel.

This coupling method was also applied to zirconacyclopentadiene 12. When 12 was treated with CuCl (2.0) equiv.) and DMPU (2.5 equiv.),<sup>9</sup> it reacted with two equivalents of alkynyl iodide to give diene-diyne compounds 13 (eq. 3). However, when the frequently used CuI/Pd/amine condition<sup>10,11</sup> was applied to diiododiene 14 and 1-hexyne, not 13b but a fulvene derivative 15<sup>12</sup> was obtained in a low yield (eq. 4).





Furthermore, a combination of two successive coupling reactions gave unsymmetrically substituted diene derivatives from symmetrical zirconacyclopentadienes. The first coupling reaction proceeded with aryl iodide in the presence of CuCl and DMPU, under the same conditions as we reported,9 to give monoarylated compound 16 as an intermediate. Interestingly, no diarylation products were obtained at this stage even when an excess amount of iodobenzene was used. Since one metal-carbon bond remained on the intermediate 16, the second coupling was successful with an alkynyl iodide to afford 3,5-dien-1-yne compound 17 in one-pot in excellent yields (eq. 5). The use of alkynyl iodide did not give iodination products of 16.



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Alkenylzirconium	$R^{4}$ $-x$	Products	Yields, % GC (Isolated)
8a	$Bu \rightleftharpoons$ -Br	Ŗr Pr	83 (76)
8a	Bu— <del>—</del> -1	Et Bu	68 <sup>b</sup>
8a	$Ph =$ -Br	Ŗ٢ Pr Et Phí	81 (50)
8b	$Ph \equiv -Br$	Ŗh Ph Ét Ph	64 (49)
9a	-Br Bu—≣	Pr P۱ Bú	91 (67)
9a	-Br Ph—	Pr Pı Ρń	87(65)
9b	Bu- <del>-</del> Br	Me <sub>3</sub> Si Bu Bú	88 (74)
10	-Br Bu—∃	Et Et Et Et Bú	66 (54)

Table 1. Formation of 1,3-Enynes<sup>a</sup>

<sup>a</sup> Conditions: 8, 9 or 10 (1.0 eq.), CuCl (1.0 eq.) and alkynyl halide (1.0 eq.), room temp., 3 h.  $b$  (E)-5-Ethyl-4-iodo-4-octene was formed in 11% yield as a by-product and was not separable from the desired product by silica gel column chromatography.

(Received in Japan 14 March 1997; revised 25 April 1997; accepted 28 April 1997)